2D-RAMAN CORRELATION SPECTROSCOPY AS A METHOD TO RECOGNIZE OF THE INTERACTION AT THE INTERFACE OF CARBON LAYER AND ALBUMIN

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Anna Kołodziej¹, Aleksandra Wesełucha-Birczyńska¹*, Paulina Moskal¹, Ewa Stodolak-Zych¹, Maria Dużyja¹, Elżbieta Długoń², Julia Sacharz³, Marta Blażewicz³

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Abstract:
In modern nanomaterial production, including those for medical purposes, carbon based materials are important, due to their inert nature and interesting properties. The essential attribute for biomaterials is their biocompatibility, which indicates way of interactions with host cells and body fluids. The aim of our work was to analyze two types of model carbon layers differing primarily in topography, and developing their interactions with blood plasma proteins. The first layer was formed of pyrolytic carbon C (CVD) and the second was constructed of multi-walled carbon nanotubes obtained by electrophoretic deposition (EPD), both set on a Ti support. The performed complex studies of carbon layers demonstrate significant dissimilarities regarding their interaction with chosen blood proteins, and points to the differences related to the origin of a protein: whether it is animal or human. However the basic examinations, such as: wettability test and nano scratch tests were not sufficient to explain the material properties. In contrast, Raman microspectroscopy thoroughly decodes the phenomena occurring at the carbon structures in contact with the selected blood proteins. The 2D correlation method selects the most intense interaction and points out the different mechanism of interactions of proteins with the nanocarbon surfaces and differentiation due to the nature of the protein and its source: animal or human. The 2D correlation of the Raman spectra of the MWCNT layer+HSA interphase proves an increase in albumin β-conformation. The presented results explain the unique properties of the C-layers (CVD) in contact with human albumin.

Keywords: Multi-Walled Carbon Nanotubes; Pyrolytic Carbon; Carbon Layers; Raman Microspectroscopy; Plasma Blood Proteins, 2D Correlation.

1. Introduction
The diverse and growing needs of modern societies can be met by the extraordinary development of nanotechnology methods that represent a unique position for various application areas and importance for economic reasons [1]. The presented research topic concerns materials for regenerative medicine due to their specificity included in the area of nanomedicine. Size of nano-additives may lead to their direct involvement in the processes and even biological structures [1-7]. Innovative biomaterials have contributed to the field of controlled drug delivery applications [8-10], cardiovascular diseases [11-13] and orthopedics [3,14], while novel materials with carbon nanotubes coatings are desirable for application as sensors and neural electrodes or as a platform for Central Nervous System diseases [15-17]. Selecting appropriate nanofiller can cause that structural, mechanical and electronic properties of nanocomposite material are shaped in a manner that would induce the desired characteristics of interaction with the biological environment [18]. So the nanoparticle by modern nanotechnology methods may be adjusted to modify, as intended, the polymer matrix [19-21], or to alter the surface of the synthetic material that comes into contact with the tissue of the living organism [11, 22-26].

Carbon materials are attractive due to their unique properties and large variety of carbon structures and nanostructures, so they can be used as a modifying particles [27]. Two types of carbon structures, pyrolytic carbon and multiwall carbon nanotubes (MWCNTs), have been chosen to identify their properties. Pyrolytic carbon can occur with different microstructures, that depend on the various forming conditions, belongs to a group of turbostratic carbons, with a structure related to graphite. However, graphite consists of carbon atoms that are covalently bonded in hexagonal arrays. These arrays are stacked and held together by weak interlayer binding, while pyrolytic carbon and other turbostratic carbons present a lack of order in the neighboring graphitic layers, so that occur wrinkles or distortions within layers. This ensures that the pyrolytic carbon has an improved durability compared to graphite [28]. Although this type of structure has been the most popular material available for artificial mechanical heart valves for about half a century there are some requirements to consider. The second type of the analyzed forms of carbon are MWCNTs, that are fibrous nanostructures [29], recently used are synthesized [27,29]. They can be also found naturally or as a byproduct of industrial processes [30,31]. The Ijima discovery caused a great interest for CNTs because of their characteristics: small size and mass, high strength, and high electrical conductivity [32].
2. Materials and Methods

2.1. Preparation of Carbon Layers on Titanium

Titanium plates (Grade 2 according to ASTM B265) in the form of discs with a diameter of 12 mm and a thickness of 0.5 mm were chosen as appropriate substrates for the deposition of carbon layers, which were: pyrolytic carbon (C (CVD)) and carbon nanotubes (MWCNTs) layer.

The first, pyrolytic C-layer was obtained in the Plasma-enhanced chemical vapor deposition process (PECVD; Elettrovara, Italy). All depositions were performed by the RF PECVD method, in which the plasma was generated by radio frequency waves of 13.56 MHz and of power 60 W. The formation of the layers was preceded by ion-etching in argon plasma during 10 minutes to eliminate of the TiOx surface layer. Then the C-layer layer was deposited at room temperature (25°C) during 30 minutes from a methane (gas flow 10 cm³/min) and argon (gas flow 75 cm³/min) mixture, while the chamber pressure was kept constant (53 Pa). Argon was used as inert gas.

The second layer of MWCNTs (#1213NMGS, Nano-structured & Amorphous Materials, Inc., USA; outside diameter 10-30 nm; core diameter 5-10 nm, length of 1-2 μm and purity >95%) was produced in the oxidizing procedure, preparation of the suspension, titanium handling and EPD set-up are presented in our previous studies [24, 25, 26].

The albumin from chicken egg white (Alb), bovine serum albumin (BSA), and human serum albumin (HSA), were purchased from Sigma-Aldrich (Poland). Both carbon coatings were incubated in 1% albumin solution for 15 minutes.

2.2 SEM Images Analysis

The morphology and chemical composition of the obtained coatings were examined using a scanning electron microscope Nova NanoSem (FEI) equipped with an adapter for EDS X-ray microanalysis (EDAX). The system was operated with 10–15 kV accelerating voltage, high vacuum mode.

2.3 Wettability Measurements

The contact angle [θ] for a liquid droplet on a tested, solid surface, was specified between the surface of the liquid and the outline of the contact surface, in the point where three phases meet: solid, liquid and gas. The contact angle measurements were taken using a direct method (DSA 10 Kruss goniometer). The tests were performed at room temperature applying the sitting drop technique (the drop of deionized water of 0.15-0.25 μl in volume). The measurements were taken five times in order to found variability and the standard deviation (SD) (that was estimated as ± 2.5%). All experiments were performed for reference and both tested surfaces and also for the selected blood proteins conditioned with tested surfaces. All tests were performed under ambient conditions.

For biomaterials the most important feature is their biofunctionality and biocompatibility, which refers to mechanical characteristics and to the interaction with host cells and fluids, respectively [33]. The biological response to the synthetic material is determined at the interface of the surface of a biomaterial and cell. One must be aware that the interface is complex, as biomolecules and synthetic material are composed of three-dimensional entities [34]. Therefore, the number of surface parameters determining the interfacial area such as: surface morphology, roughness (micro and nano), wettability, and the degree of crystallinity are parameters that result from the surface properties and significantly affect the biological properties of the material [25, 26, 35-37].

Research on the adsorption of protein to synthetic material is considered as method to assess biomaterials quality, not only that in blood-contacting applications. It is considered that the adsorption of selected proteins, may reveal the specific biological properties of the examined nanomaterials. Usually albumin is selected to study its interaction with the surface of the nanomaterial. Albumin is the blood plasma protein of the next highest abundance after hemoglobin, which is involved in transport and storage and regarded as an inhibitor of blood clotting [38].

The commonly used are proteins of animal origin, which are considered to be equivalents of human protein. However, our earlier studies have shown that animal equivalents differently affect the surface of a carbon material than human proteins [25, 26]. Therefore, modeling interactions with proteins of animal origin can be questioned in some cases. As target proteins, albumin from chicken egg white (Alb), bovine serum albumin (BSA) and human serum albumin (HSA), were employed to check their influence on the surface during performed research.

Conventional methods of material engineering such as wetting angle measurement and electron microscopic techniques (SEM) were employed to obtain the characteristics of the materials surface. The mechanical properties of the protein layers on the materials were tested with the Nano Scratch Tester. This method was applied to study the adhesion and scratch resistance of model coatings incubated with selected proteins. Then, materials were tested using Raman spectroscopy. The Raman spectroscopy was selected as an important spectroscopic technique to test short-range ordering. It has been used to describe the structure of two reference carbon layers and then the effect of adsorption on these layers of selected plasma proteins. Finally, a 2D Raman correlation spectroscopy was applied to the collected Raman spectra and allowed for the resolution of the phenomena occurring at the interphase, the carbon layer surface in contact with albumin, the most abundant blood proteins. In this mathematical analysis as an external perturbation the spatial position, in which the spectrum was measured, was taken into account.
2.4. The Nano Scratch Test

All tests were carried on a platform with the NST (Nano Scratch Tester) head made by CSM Instruments SA (currently Anton Paar TríTec) (Switzerland). The parameters, while testing, were set as follows: the load \( F_n \) increased linearly from 0.1 to 5 mN on the 3 mm distance, the speed of loading was set for 10 mm/min and the Rockwell certified indenter radius was equal to 2 μm.

Two nanocomposite layers were analyzed: the C-layer grown in the CVD process and the MWCNTs deposited in the Ti support, after incubation with chicken egg white albumin and human serum albumin (HSA).

2.5. Raman microspectroscopy

A Renishaw inVia spectrometer, connected to a Leica microscope, was used for the measurements of the Raman spectra. The beam from a 514.5 nm Ar+ ion Modu Laser was focused on the samples by 100x magnifying, a high numerical aperture (NA = 0.9) top-class Leica objective for standard applications. Laser power was kept sufficient low, c.a. 1-3 mW at the sample, to ensure minimum disturbances to the samples.

2.6. 2D Raman Correlation

The generalized 2D correlation analysis based on the Noda method [39-41] was performed using Raman spectra as an input data for generating the correlation maps. The spatial position was regarded as an external perturbation [42]. The five points, morphologically similar were measured for each sample, and they were regarded as a dynamical spectra in the 2D correlation. 2Dshige, v.1.3 software was employed [43].

3. Results and Discussion

3.1. SEM Images of Carbon Layers

SEM investigations indicate dissimilarities in the topography of both materials. Two types of carbon structures are different primarily in the topography. The first coating was a layer formed of pyrolytic carbon (CVD) (Fig.1A) and the second was constructed of multi-walled carbon nanotubes obtained by electrophoretic deposition (EPD) (Fig. 1B), both set on a Ti support. A slight, intentional difference in the magnification of both images allows to see the specificity of the structure of micro- and nano-layer of C (CVD) and MWCNTs, respectively.

3.2. Wettability of Carbon Layers

The surfaces wettability was analyzed by the static sessile drop method. The contact angles of water droplets on the top face of the reference C (CVD) layer are 82.2 ± 2.8°, respectively (Figure 2A). The C (CVD) layer does not change the specificity of the surface with respect to the titanium substrate [26]. The difference in the contact angles for C (CVD) incubated with Alb (78.4 ± 2.3°) and HSA (84.5 ± 1.9°) with comparison to a reference the C (CVD) layer is not so significant, it fits within the limit of 5% (Figure 2A). However, the C (CVD) incubation in BSA leads to the creation of a film with hydrophobic characteristics and contact angle of 112.0 ± 6.7°.

The contact angles on the top face of the MWCNTs nanocomposite layer is 25.0 ± 0.9° (Figure 2B). This value implies a hydrophilic character of the surface of the MWCNTs coating. The Alb and HSA form a layer having a wettability 62.6 ± 1.3° and 57.3 ± 0.3°, respectively. The BSA conditioned MWCNTs nanocarbon layer reaches the highest contact angle of 77.1 ± 1.3°.
3.3. Nano Scratch Test

The performed test consisted of three steps. In the first step the profile of the sample (Pre Scan) was collected. In the second phase the indenter was pressed into the sample with linearly increasing load and moving along the sample at a defined length to scratch off the coating (test phase). During the third step the profile inside scratch (Post Scan) was accumulated. After the performed scratch the panoramic photograph is taken and the obtained features are analyzed.

HSA has the best adhesion to the C (CVD) layer (Table 1). This coating did not break during the performed test and the Ti substrate is not visible. It can be observed that the C (CVD) layer with a thin HSA sheet looks like a “pressed and smeared” during the test and therefore presents the best adhesion to the Ti base. In addition, violation of coating, chipping or cracking up to the 5mN load is not observed.

Tab. 1. Summary of the Scratch test Parameters (linear scratch; load range 0.1-5 mN; loading rate 10 mN/min).

<table>
<thead>
<tr>
<th>Sample</th>
<th>critical load [mN]</th>
</tr>
</thead>
<tbody>
<tr>
<td>C (CVD) + Alb</td>
<td>coating failed at 2.16 ± 0.20</td>
</tr>
<tr>
<td>C (CVD) + HSA</td>
<td>coating is not destroyed</td>
</tr>
<tr>
<td>MWCNTs + Alb</td>
<td>coating failed at 1.63 ± 0.10</td>
</tr>
<tr>
<td>MWCNTs + HSA</td>
<td>coating failed at 2.17 ± 0.16</td>
</tr>
</tbody>
</table>

For the MWCNTs nanolayer the HSA film is de-laminated with a load of about 2-2.5 mN. It is observed that the entire carbon nanolayer along with the thin film of protein is detached from the Ti substrate.

Adhesion of the Alb to the carbon C (CVD) layer is better, but the difference is small, the surface is scratched with a load of about 2.16 mN. The surface of the Alb on the MWCNTs nanolayer is scratched off easily, the visible surface of the Ti substrate is noticeable from the start of the test which means that the critical load for this layer is below 1.63 mN.

3.4. Raman Microspectroscopy Characterization of Carbon Layers and their Interaction with Selected Blood Proteins

The Raman band positions and assignments of the reference carbon layers excited by the 514.5 nm laser line are collected in Table 2. The first, the C (CVD) layer is formed of pyrolytic carbon, which is an anisotropic material. Therefore, only the main G- and D-bands are observed, what proves the graphite-type arrangement in this coating (Figure 3A). For the second, MWCNTs nano-layer, in addition to G- and D- also characteristic 2D, D’ and D+D’ bands are visible confirming more complex organization in this coating (Figure 3B).

Tab. 2. Observed Raman bands [cm⁻¹] and their assignments for carbon coatings excited with 514.5 nm laser line.

<table>
<thead>
<tr>
<th>Sample</th>
<th>Peak position [cm⁻¹]</th>
<th>Assignment [20-28, 44-47]</th>
</tr>
</thead>
<tbody>
<tr>
<td>C (CVD)</td>
<td>1349</td>
<td>D-mode; breathing mode that requires a defect for its activation</td>
</tr>
<tr>
<td>MWCNTs/ (EPD)</td>
<td>1364</td>
<td></td>
</tr>
<tr>
<td>C (CVD)</td>
<td>1594</td>
<td>G-mode; E₂g mode at the Brillouin zone center</td>
</tr>
<tr>
<td>MWCNTs (EPD)</td>
<td>1595</td>
<td></td>
</tr>
<tr>
<td>MWCNTs (EPD)</td>
<td>1633</td>
<td>D’-mode; effect of double resonance as an intravalley process</td>
</tr>
<tr>
<td>MWCNTs (EPD)</td>
<td>2713</td>
<td>2D (G’); second order of the D peak</td>
</tr>
<tr>
<td>MWCNTs (EPD)</td>
<td>2956</td>
<td>D+D’; combination of phonons with different momenta, requires a defect for theirs activation</td>
</tr>
</tbody>
</table>

The G-band at ca.1580 cm⁻¹ is typical for the sp² carbon materials and is assigned to the high frequency E₂g optical phonon [44]. The G-band position for the C (CVD) differs from that for the MWCNTs (EPD) layer, which is more heterogenous [45, 46].

Due to the excellent sensing properties of the carbon particle, the interaction with proteins can be observed and determined. The interaction occurs in the interphase region, between the two phases, and depends on both of them, on the type of carbon material and on protein (Fig. 4). The noticed shift of the G-band position shows the identifiable type of interaction in relation, first of all, to the reference coating. Furthermore, the positions of the G-band for the animal albumins, Alb and BSA, differ from that of HSA, what confirms that interaction occurs between the thin protein layer and the carbon coating as well as their specificity for different types of albumin.

Therefore, you need to consider what properties of the carbon layers are crucial in the application you are working on.
Fig. 4. Position of the G-band for the reference carbon layers (A) C-(CVD), and (B) MWCNTs (EPD), and after incubation with the selected proteins; 514.5 nm excitation line.

3.5. 2D Raman Correlation Method

Two-dimensional correlation uses mathematical formalism to obtain two-dimensional correlation spectra from any transient or time-resolved spectra of an arbitrary waveform [39, 40]. The experimental approach takes into account that an external perturbation, applied to a studied system, selectively excites various chemical constituents of the system. A 2D experiment in optical spectroscopy is performed by introducing a relatively slow external perturbation applied to the system of interest [41]. Spectral changes that are noticed under certain dynamic perturbation are the variation of intensities, shifts of spectral band positions, and alteration in the shape of peaks. These fluctuation of spectral signals are transformed into two-dimensional spectra by using a correlation method formalism. The type of physical information enclosed in a dynamic spectrum, is determined by the selection of perturbation method and electromagnetic probe. Therefore, 2D spectra obtained by 2D correlation method can highlight valuable information often hidden in the original time-resolved spectra.

3.6. 2D Raman Correlation Spectroscopy Characterization of Carbon Layers Interaction with Selected Blood Proteins

Carbon materials are characterized with a very high polarizability, therefore Raman spectroscopy is the method of choice for their analysis. However, the studied interaction occurs primarily in the interphase area, so not only carbon material, but also the complex biological structures of albumin are involved in the interaction. The most interesting would be to analyze the spectra in the vibration range of amide I, which is also the region of appearance of the D- and G-band of carbon layers. However, the Raman signal from the carbon component on the phase boundary might cover the spectrum of the protein, due to the high polarizability of sp² system. Hence, a two-dimensional correlation analysis was used to control changes in the structure of the proteins on the interaction with the carbon layers in order to decode the relations hidden in the Raman spectra [40, 42]. In the performed 2D correlation spectroscopy analysis the variable intensities were linked with a location on a sample characterizing the respective protein film - carbon coating interactions. The 2D spectrum indicates the clear differentiation between the origins of the Raman spectral signals [40].

Synchronous signal fluctuations indicate a common chemical constituent at ca. 1588 and 1345 cm⁻¹ and also at ca. 1613 and 1313 cm⁻¹ originating from the graphite G and D-band components of the studied coating for the C (CVD) and MWCNTs layer, respectively.

The nonsynchronous signal fluctuations indicate chemically dissimilar components, thus the map pattern differs for the two carbon coatings and additionally animal albumins are clearly distinguishable from that of human (Figs. 5 & 6). The asynchronous correlation map for the C (CVD) + Alb layer in the 1720–1530 cm⁻¹ range shows intensive negative cross-peaks originating from the amide I component bands at 1650 cm⁻¹ (α-helix conf.), 1665 cm⁻¹ (β-sheet conf.) and 1682 cm⁻¹ (β-turn conf.) with the 1597 cm⁻¹ band due to the G-band of the carbon layer. Otherwise it is for the C (CVD)+ BSA layer, there is an intensive positive band in a different location at (1600, 1552 cm⁻¹) that arises from the carbon G-band and Glu vibrations. The C (CVD)+ HSA layer presents a positive asynchronous peak derived of 1635 cm⁻¹ (Trp, Arg, His) and 1652 cm⁻¹ (α-helix conf.) and 1661 cm⁻¹ (β-sheet conf.) of amide I and at ca. 1594 cm⁻¹ of the carbon layer G-band.

Fig. 5. Asynchronous 2D correlation Raman spectra of the C (CVD) sample incubated with: (A) chicken egg white albumin (Alb); (B) bovine serum albumin (BSA); (C) human serum albumin (HSA); in the wavenumber range of 1720-1530 cm⁻¹; the red and blue color represent positive and negative cross peaks, respectively.

A different pattern is observed for the 2D asynchronous maps for the second type carbon layer. The MWCNTs + Alb present a positive asynchronous correlation cross-peak at 1600 cm⁻¹ of protein aromatic ring vibrations and 1622 cm⁻¹ of Tyr with 1578 cm⁻¹ of carbon nanotubes G vibration. Another tested system MWCNTs +BSA shows intensive negative cross peak – (1630,1602 cm⁻¹) owed to His, Tyr and (G') carbon band and –(1582,1603 cm⁻¹) originating from the G'-band of the MWCNTs and the Phe albumin vibrations.
The MWCNTs+ HSA layer gives a negative asynchronous peak -1618,1602 cm\(^{-1}\) due to the Tyr and G\(^{+}\) carbon band. The other cross-peak originated of the amide I band of 1659 cm\(^{-1}\) (β-sheet conf.) and 1687 cm\(^{-1}\) (β-turn conf.) with 1601 cm\(^{-1}\) (G\(^{+}\)) carbon nanotube vibrations.

The calculated asynchronous cross-peaks are enclosed in Table 3. The 2D correlation spectroscopy allows to differentiate the adhesion specificity of the selected blood protein to the studied model, carbon layers.

### 4. Summary

The results of the performed complex studies of the two types of model carbon coatings display significant dissimilarities regarding their interaction with the chosen blood proteins but also the difference is related to the origin of a protein: whether it is animal or human. Both of the studied carbon layers were incubated with the selected albums, and the interaction between these two materials is visible by the variation of the contact angle (Fig. 2). They substantially differ in their surface image, the nanotubes form an isotropic fibrous system with a characteristic nano topography while the C (CVD) pyrolytic surface is smoother (Fig. 2B).

For both types of the studied carbon layers a similar sequence of changes reflecting the occurring interaction with the albums, which can be estimated by measuring contact angle. For each of the albumin, these differences are clearly marked, and a remarkably different contact angle is noticed for BSA (Fig. 2 A and B). The CVD layer, which is known to be antithrombogenic, is characterized by high adhesion of protein to the surface while in other cases, the protein layer weakly adheres to the substrate (Table 1).

The results of Raman spectroscopy indicate that carbon layers interact differently with the selected blood proteins, as indicated by the parameters determined from the Raman spectra, e.g. the position of the characteristic G-bands (Fig. 3). This parameter allows to uncover the type of interactions and their extent.

The 2D asynchronous maps offer the possibility of determining the type of protein interaction with the surface of the carbon layer Considering only the most intense cross-peaks you will notice that Alb adopts the structure with a comparable contribution of the α-helix, β-sheet and a sizable portion of β-turn conformation while signals from the individual amino acids Phe, His, Tyr are observed for the MWCNTs layer (Fig. 5A & 6A) [48-50]. For the BSA there is a correlation signal from the individual amino acids: Asp, Glu and Phe, His for C (CVD) and MWCNTs layer, respectively (Fig. 5B & 6B ). [47-49]. The 2D correlation spectroscopy does not provide evidence that the secondary structure is mainly α-helix (50-60% in its native state) [49,51,52]. HSA acts in an exceptional way with a synthetic nanomaterial, showing equal participation conformation α-helix and also α-helix and

<table>
<thead>
<tr>
<th>2D asynchronous cross-peaks</th>
<th>C (CVD) + Alb</th>
<th>MWCNTs + Alb</th>
</tr>
</thead>
<tbody>
<tr>
<td>assignment</td>
<td>cross-peaks</td>
<td>assignment</td>
</tr>
<tr>
<td>amide I, α-helix conf.</td>
<td>-(1650,1597)</td>
<td>G-band Phe, His</td>
</tr>
<tr>
<td>amide I, β-sheet conf.</td>
<td>-(1665,1597)</td>
<td>G-band Tyr</td>
</tr>
<tr>
<td>amide I, β-turn conf.</td>
<td>-(1682,1597)</td>
<td>G-band</td>
</tr>
</tbody>
</table>

| Observed significant asynchronous 2D correlation cross-peaks and their assignments for the C (CVD) and MWCNTs coating incubated in Alb, BSA, HSA in the wavenumber ranges of 1750–1500 cm\(^{-1}\) [20-28, 44-47]. |
|-------------------------------------------------|-------------------------------------------------|
| 2D asynchronous cross-peaks | C (CVD)+ Alb | MWCNTs + Alb |
| assignment                  | cross-peaks | assignment | cross-peaks | assignment |
| amide I, α-helix conf.      | -(1598,1552) | C (CVD)+ HSA | +(1598,1552) | G-band    |
| Glu                         | +(1613,1558) | MWCNTs + HSA | + (1613,1558) | G- band    |
| D’-band                     | +(1635,1601) | G-band Tyr | +(1630,1603) | G- band    |
| C (CVD)+HSA                 |             | amide I, α-helix conf. | +(1652,1601) | G-band |
| D’-band                     | +(1635,1594) | G-band       | -(1618,1602) | G- band    |
| amide I, α-helix conf.      | + (1652,1601) | G-band amide I, β-turn conf. | -(1687,1601) | G- band    |

**Fig. 6.** Asynchronous 2D correlation Raman spectra of the MWCNTs (EPD) sample incubated with: (A) chicken egg white albumin (Alb); (B) bovine serum albumin (BSA); (C) human serum albumin (HSA); in the wavenumber range of 1720-1530 cm\(^{-1}\). The red and blue color represent positive and negative cross peaks, respectively.
b-turn in addition to the Tyr signal for the C (CVD) and MWCNTs layer, respectively [49,53]. But, the vibrations of the amide I are ahead of the changes in the carbon nanolayer for the C (CVD). The MWCNTs layer shows the opposite order of events (Fig. 5C & 6C). The 2D analysis shows that the amide I bond is modified by the aromatic MWCNTs structure while the interaction on the interphase occurs, what was also noticed for the SWCNTs [54].

5. Conclusion

The albumin conformation is different for the studied surfaces, the amide I band maximum observed for the C (CVD) pyrolytic carbon layer shifts toward the higher vibrations for the MWCNT coating confirming an increase in the amide I of β-conformation. The conducted research and spectroscopic characteristics of the studied surfaces topography as a key element in the synthetic surface with blood protein interaction and allow for the explanation of the nature of this process in relation to the type of protein.

The phenomena occurring on the surface of the C-pyrolytic carbon with contact with human albumin have a different character than those observed in other cases. It can therefore be assumed that these phenomena, leading to the conformational changes in HSA and strong adhesion, indicates non-thrombogenic characteristic of this type surface coatings.

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